Karabatak N et al. JOTCSA. 2024; 11(3): 1181-1196 **RESEARCH ARTICLE**

Development of the Nanoemulsion Formulation Containing Ylang Ylang Essential Oil for Topical Applications, and Evaluation of Its *In Vitro* **Cytotoxicity as well as ADMET Profile**

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Abstract: Ultraviolet (UV) rays damage DNA, causing adverse effects such as photoaging and cancer on the skin. For the well-being of individuals, there is a need to develop innovative skin products with high effectiveness using protective and therapeutic agents. In our study, a nanoemulsion (NE) formulation containing Ylang-ylang essential oil (YO), which has many biological active properties such as antimicrobial, antioxidant, anti-inflammatory, and anticancer, was produced by the ultrasonic emulsification method and characterized. The thermodynamic stability was evaluated, and its release profile determined the dialysis membrane technique. The cytotoxic effect of YO-NE was examined with the *in vitro* method on the HacaT cell line using the MTT method and *in silico* method using the ADMET profile. Dynamic light scattering (DLS) results showed that the average droplet size of the YO-NE formulation was 184.1±2.307 nm, the polydispersity index (PdI) was 0.151 ± 0.006 , and the Zeta potential (ζ) -10.8 \pm 0.400 mV. As a result of release studies, it was observed that 99.98± 1.00% of YO release from NE occurred within 5 hours. Based on the thermodynamic stability test results, it was determined that the developed formulation did not show sedimentation or phase separation. Cytotoxicity results revealed that the YO-NE formulation was safe. All the results indicated that the YO-NE formulation might be considered a non-toxic product candidate with physicochemical properties suitable for topical use.

Keywords: Nanoemulsion (NE), Ylang-ylang essential oil (YO), Cytotoxicity, Topical application.

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1. INTRODUCTION

Skin, the largest body organ, plays a critical role in maintaining homeostasis and creates a protective barrier between the environment and the body (1). However, exposure of this barrier to excessive sunlight may disrupt the barrier function of the skin (2). Exposure to ultraviolet (UV) radiation can lead to the overproduction of reactive oxygen species (ROS) in skin cells, DNA damage, lipid peroxidation, protein modification, and cell apoptosis (3). These factors can lead to skin problems such as photoaging, wrinkles, inflammation, and even skin cancer (4). Plants or bioactive components obtained from plants can be used to overcome these problems (5-7).

Plants and active substances derived from plants are widely used in producing cosmetic materials for protection from UV rays, antioxidants, cell renewal and minimizing the effects of photoaging (8). Essential oils (EOs) from plants, their usage in the plant defence mechanisms against different parasite species and infections is used in various medical approaches with biochemical effects such as antiviral, anticancer, antibacterial, and antiinflammatory activities (9). Ylang-ylang, also commonly known as *Cananga odorata*, is a fastgrowing tree commonly found in tropical Asian countries such as Malaysia, Philippines, Indonesia and some islands of the Indian Ocean. Ylang-ylang EOs are currently widely used in the food, perfume,

cosmetics industry and aromatherapy to treat many diseases such as asthma, fever, inflammation, wounds, microbial infections, colds, etc (10,11). The Ylang-ylang plant and especially the EOs obtained from it contain antioxidant components. The antioxidant effects of EO obtained from ylang-ylang help to reduce oxidative stress and positively support body health by fighting free radicals that cause cellular damage (12).

Although EOs have many biological activities, their clinical applications are limited due to their hydrophobic structure and poor stability in different environmental conditions such as air, light, humidity, and high temperature. To overcome the limitations in the clinical use of EOs, they are prepared in nanosized and suitable dosage forms for the application method (13). The development of innovative formulations in the field of nanomedicine has enabled increasing the therapeutic efficacy and reducing the toxicity of natural compounds and their bioactive components (14-16).

Among nanoformulations as nanoemulsions (NEs), frequently preferred for topical use, are translucent and/or transparent emulsions characterized by nanosized droplets (17). When used as a topical carrier, small droplets have a large surface-to-volume ratio, allowing the activated compound to spread easily into the skin and provide high absorption. In addition, nanoemulsions can increase the solubility of lipophilic compounds and change the skin's diffusion barrier depending on the nanoemulsion's composition. Thus, they may enable the drug to penetrate better into the skin layers (18,19). The nanometer size of NE improves its transmission target and specificity, making it more perfect and effective than pure EO (20). NE formulations are safe for human health. They increase the solubility, bioabsorption, biomembrane permeability, and bioavailability of poorly soluble active substances. Additionally, NEs are biocompatible, biodegradable, and do not have mutagenic effects. Thanks to their controlled release properties, NEs minimize the toxicity of the active ingredient and contribute to a better therapeutic effect (21,22). Additionally, NEs contain fewer surfactants than microemulsions, and these surfactants are environmentally friendly, costeffective, and economically viable (20).

In this study, YO-NE nanoformulation was prepared using the ultrasonic emulsification method and then characterized by different techniques. DLS was used to determine parameters such as average droplet size, ζ potential, and PdI. The thermodynamic (centrifugation and thermal stress test) and physicochemical stability of the YO-NE formulation were evaluated. The release profile of the YO-NE formulation was determined with a UV-Vis spectrophotometer using the dialysis membrane method. The cytotoxic activity of YO-NE was evaluated by the MTT method using the HacaT cell line. Finally, the ADMET profile of YO was considered.

2. EXPERIMENTAL SECTION

2.1. Material

Ylang-ylang oil was purchased from Aksuvital, Ethanol, Kolliphore® P-188, DL-alpha tocopherol acetate from Sigma-Aldrich; undecyl alcohol and isodecyl neopentanoate were purchased from Schülke. Caprylic/capric triglyceride was purchased from Gattefosse. The HaCaT cell line used in the MTT assay was purchased from Thermo Fisher Scientific, MTT from Biomatics, and trypsin/EDTA (0.25%) was purchased from Gibco. Fetal Bovine Serum (FBS) and penicillin-streptomycin Solution were purchased from Biological Industries.

2.2. Method

2.2.1. Development of the YO-NE

YO-NE formulation, in which the aqueous phase is a continuous phase (O/W), was prepared using the ultrasonic emulsification method (23). The water phase of NE was obtained by dissolving 7.5% Kolliphore® P-188 in water. The oil phase was formed by the addition of 1% undecyl alcohol, 1.25% transcutol, 1.5% isodecyl neopentanoate, 10% caprylic capric triglyceride, 0.2% DL-alphatocopherol acetate, 3% Labrafil and 0.5% YO. These phases were prepared separately, and then the water phase was added to the oil phase. The premixing step was performed using a homogenizer (Witeg, Germany) at 8100 rpm for 5 minutes. After pre-mixing, the emulsion was ultrasonicated at 50% amplitude for 20 minutes using a 20 kHz and 750 W sonicator (Ultrasonics, USA).

2.2.2. Analysis of droplet size, PdI and ζ potential of YO-NE

The parameters such as PdI, average droplet size, and Zeta potential (ζ) values of YO-NE were measured using Zeta Sizer Nano ZS (Malvern Instruments, UK). All measurements were performed at 25°C. Specimens were diluted as 1:100 in sterile water and carried out in triplicate (24).

2.2.3. Analysis of pH and electrical conductivity of YO-NE

A pH meter (Ohaus® STARTER 3100M) with a conductivity probe was used to determine the electrical conductivity and pH of the YO-NE formulation. All analyses were performed in triplicate at 25°C.

2.2.4. Active ingredient content analysis

Active ingredient content analysis is used to assess the stability of drugs in pharmaceutical formulation. In this experiment, the YO-NE formulation (250 µL) was dissolved in 10 mL ethanol and sonicated in an ultrasonic bath for 30 min (25). The YO content in the sample was then determined using the equation for the spectrophotometric analysis curve of YO.

2.2.5. Morphology analysis

The morphology of YO-NEs was examined by transmission electron microscopy (TEM) (JEOL TEM 1400 Plus). A carbon-coated grid was used in the analyses. A sufficient amount of the formulation was placed on the grid, and images were displayed under a voltage of 80 kV (26).

2.2.6. YO-NE accelerated stability tests

Following the preparation of the YO-NE formulation accelerated thermodynamic stability tests consisting of centrifugation and thermal stress tests were carried out within 24 hours. 0.5 g of the YO-NE formulation was centrifuged at 4500 rpm for 30 minutes at 25±1°C. Macroscopically, it was evaluated for any phase separation or turbidity (27). For thermal stability testing, 0.5 g of YO-NE formulation was carried out in a water bath at 40- 80°C with temperature increments of 5°C each. The thermal stability test of the YO-NE formulation was evaluated in terms of organoleptic features organoleptic aspects such as color, odor, texture and phase separation.

2.2.7. Calibration curve

Seven different YO concentrations (2.34, 4.68, 9.36, 18.75, 37.5, 75 and 150 µg/mL) were prepared in ethanol. Then, the absorbance values of these samples were measured with a UV-Vis Spectrophotometer (Shimadzu, Japan) at 283.2 nm, and the calibration curve was drawn (28). This curve was used to determine the amount of YO in the release study of YO-NEs.

2.2.8. Release study

Since the YO-NE formulation is intended for topical application, the release study was performed at skin pH 5.5 and temperature (32°C) (29,30). 1 g of YO-NE was added to the dialysis capsule, and 1 mL of release medium was taken at specified time intervals from a water bath shaking at 120 rpm and added with an equal volume of fresh release medium. The amount of YO released was determined by analyzing the absorbance of the samples with a UV-Vis spectrometer at 283.2 nm. YO release (%) was calculated using Equation (1).

$$
Release\% = \frac{Released\,YO}{Total\,YO} \bullet 100\tag{1}
$$

2.2.9. Preparation of cell culture

The cytotoxic effect of YO and YO-NE on HaCaT cells was evaluated using MTT assay. HaCaT cell line was incubated in Dulbecco's Modified Eagle's Medium (DMEM) medium containing 100 U/mL penicillin-100 g/mL streptomycin antibiotic and 10% Fetal Bovine Serum (FBS) in a 5% $CO₂$ incubator at 37°C until 80% confluent. The proliferating cells were washed with 5 mL sterile phosphate buffer solution (PBS). To lift off the cells from the flask, 1 mL of trypsin-EDTA was added and incubated for 5 minutes. To neutralize the effect of trypsin, 5 mL of culture medium was added to the cells and the cells were centrifuged at 123 g for 5 minutes. The supernatant was discarded, and the pellet was dissolved in 1 mL of medium and counted on a thoma slide (31).

2.2.10. Cell viability test (MTT)

The *in vitro* safety of YO and YO-NE was evaluated by MTT test using the HaCaT cell line. In this method, cells were seeded in a 96-well plate $(1x10⁴/well)$ and incubated for 24 hours at 37ºC in a humidified incubator with 5% $CO₂$. After incubation, five different concentrations of YO and YO-NE (0.25, 0.5, 1, 3 and 5 mg/mL) were added to the wells and

incubated for 24 hours. In this test system, 0.1% DMSO was used to prepare concentrations of YO. DMEM was used to prepare the concentrations of NEs. For YO, DMEM containing 0.1% DMSO was used as a control, while for YO-NE, only DMEM was used. Ten μL of MTT solution was added to each well and incubated for 4 hours. The medium was carefully removed, and the colored formazan crystals were dissolved in 100 μL of dimethyl sulfoxide (DMSO). The absorbance given by the cells in the plates was measured at 570 nm using a microplate reader. YO and YO-NE untreated control value was expressed as 100% cell viability. Equation (2) was used to determine cell viability.

Cell viability = (OD_{570} of sample / OD_{570} of control) \times 100 (2)

2.2.11. Physiochemical and pharmacokinetic analyses of linalool and germacrene-D

Linalool is found in many aromatic plant families, such as Rutaceae and Lamiaceae. Linalool is a naturally occurring acyclic monoterpene alcohol. Numerous traditional and edible plants, including coriander, peppermint and cinnamon, produce essential oils containing linalool (32). The Food and Drug Administration (FDA) has deemed linalool as GRAS safe (GRAS) as a synthetic flavoring agent as well as an excipient in foods for humans (21 CFR 182.60) and as an ingredient in animal drugs and foods (21 CFR 582.60) (33). The chiral hydrocarbon germacrene-D is a widely available plant component that is recognized as a critical intermediate in the biosynthesis of many sesquiterpenes (34). These sesquiterpene hydrocarbons are biosynthetic precursors of various (often oxidized) structures from which humans have exploited many compounds with significant medicinal and other bioactivities (35).

Kalagatur et al. (2018) (11) characterised ylangylang oil by GC-MS and reported that the main compounds in the oil were linalool (29.15%), germacrene-D (11.82%) and thymol (8.45%). In this study, physicochemical and pharmacokinetic analyses were carried out due to the high amounts of linalool and germacrene-D compounds in ylangylang oil.

Drug susceptibility characteristics were analyzed using SwissADME (absorption, distribution, metabolism and excretion) and the PkCSM tool to predict crucial pharmacokinetic properties such as drug candidate molecules' absorption, distribution, metabolism, excretion, and toxicity (ADMET). The canonical simplified molecular input line entry system (SMILES) format of the compound was retrieved from PubChem (https://pubchem.ncbi.nlm.nih.gov/) and entered into SwissADME and PkCSM (http://www.swissadme.ch/)

(https://biosig.lab.uq.edu.au/pkcsm/) has been sent (36,37). Drug susceptibility is determined by Lipinski's rule of five (RO5) (38).

2.2.12. Estimation of biological activity

The evaluation of the overall biological potential of the compounds was performed using PASS (http://195.178.207.233/PASS/index.html). This software estimates the predicted activity spectrum of a compound with probability (Pa = probability of 'activity' and Pi = probability of 'inactivity'). The cutoff value was set as Pa≥0.7 (39). This method is based on the training set's structural activity relationship (SAR) analysis, which contains more than 205,000 compounds exhibiting more than 3750 biological activities. Compounds that show a probability that the Pa value is higher than the Pi value are those considered to be possible for a specific pharmacological activity (40).

3. RESULTS AND DISCUSSION

3.1. Characterization of YO-NE

The ultrasonic homogenization method is a fast and effective method to create stable NEs with appropriate droplet sizes and low PdI (41,42). In this study, NE formulation was synthesized with ultrasonic homogenization, followed by characterization and stability studies.

DLS, one of the most preferred methods, was employed to determine the hydrodynamic size, PdI, and ζ potential value of the synthesized NE (43,44). The results were given in Figure 1 and Figure 2. The average droplet size of the empty NE was 183.2±1.498 nm. However, the average droplet size of YO-NE was 187±2.307 nm. The most important advantage of NEs over emulsions is that they are nanosystems with small droplet sizes of 20-200 nm (45). The small droplet size of NEs provides many advantages, such as overcoming difficulties, separation, flocculation and coalescence due to gravity (46,47). Many studies have reported that droplet sizes between 100-200 nm may benefit NEs in topical applications due to their excellent penetrability and controlled release capabilities (48,49). Based on the average droplet size data, the

PdI value is essential for the homogeneity of droplet sizes in NE and the stability of the formulation. A PdI value of <0.1 means a narrow size distribution of droplets, and this is the indication of the formulation homogeneity. A PdI value of >0.2 indicates a wide dimensional distribution of droplets, i.e., heterogeneity of the formulation (50). It was determined that the PdI value of empty NE was 0.190±0.020 and the PdI value of YO-NE was 0.151 ± 0.006 . The results showed that the droplets formed have a tight size range, i.e., homogeneous.

Ozdemir et al. (2023) (51) found that the average droplet size of the NE formulation (F2) containing etodolac was 163.5±2.2, PdI 0.141±0.02, zeta potential- 33.1±1.7. Gündel et al. (2020) (52) found the average droplet size of NE formulation containing eucalyptus as 68±0.15, PdI 0.18±0.01, zeta potential -9.09±0.65. Our results coincided with those presented in previous studies in the literature.

ζ potential is used to calculate the charge on the surface of NE droplets and to indirectly determine the electric charge of particles in a heterogeneous system (53). The minus charge in the ζ potential value is due to the carboxylic acid groups of fatty acids used in the formulation of fatty acids (54). The ζ potential of empty NE was found to be -10.7±0.252 mV and the ζ potential of YO-NE was found to be - 10.8±0.400 mV.

Kildaci et al. (2021) (55) obtained a stable LSO-NE for topical application, an acceptable mean droplet size of 90.61 ± 0.94 , a PdI value of 0.15 ± 0.008 and a zeta potential value of 9.64±0.55 mV. Kilinc et al. (2022) (56) obtained a stable CA-NE for topical application and found an average droplet size of 120.4±6.39 nm, PdI value of 0.180±0.018, and zeta value of 11.5±1.15 mV. This study's results were similar to those presented in previous studies in the literature.

Figure 1: DLS analyses of the empty NE. **(a)** Average droplet size plot and PdI, **(b)** ζ potential graph.

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Figure 2: DLS analyses of YO-NE **(a)** Average droplet size plot and PdI, **(b)** ζ potantial graph.

3.2. pH and Conductivity

The pH value is crucial for assessing the stability of nanoemulsions. pH changes in the formulation suggest the possibility of chemical reactions leading to problems in the stability and quality of the final product (57). The pH value of the skin is approximately 5.5 and generally a pH in the range of 4.0 to 7.0 is appropriate for topical applications. In conclusion, according to the data obtained from our study, the pH value of YO-NE is 5.83, which makes it appropriate for topical application (Table 1). Rashid et al. (2021) (58) determined the pH value of methotrexate-loaded nanoemulsion as 5.81±0.22, which is an acceptable value for application as transdermal systems since it is within the specified range scale of the physiological pH of the skin. Hammodi et al. (2020) (59) found the pH value (5.96±0.025) of letrozole loaded NE formulation to be in the acceptable range for topical use. Our results are similar to those results presented in previous studies in the literature.

Conductivity is the measurement of the amount of free ions and water present in the solution and the determination of its response. The O/W formulation is suitable for use in the cosmetic industry as its structure is less oily after topical application (60). This parameter allows the determination of the kind of NE prepared. The high conductivity of YO-NE (92.63 µS/cm) indicates that the aqueous phase is a continuous phase and the nanoemulsion formed is an oil-in-water nanoemulsion (O/W). In this study, YO-NE formulations' pH and conductivity values are higher than empty-NE (Table 1). Wang et al., (2020) (61) reported the effect of pH on both the hydrodynamic diameter of droplets and the conductivity of NEs. In this effect, the hydrodynamic diameter of the droplets gradually increases with increasing pH. Regulation of pH changes the ionic strength and, therefore, affects the droplet size of NEs. As the pH value increases, the conductivity of NEs gradually increases, which means that the ionic strength increases simultaneously. These results explain why YO-NEs have higher pH and conductivity values than empty-NEs.

Table 1: The pH and conductivity values of NEs. (Data are presented as mean±SD).

Formulation	pН	Conductivity (µS/cm)
Empty-NE	4.48 ± 0.02	87.33±0.65
YO-NF	5.83 ± 0.02	92.63 ± 5.90

3.3. Active Ingredient Content Analysis

Active substance content analysis is critical for any dosage form. The amount of active ingredient in the product should not deviate from the amount specified on the label beyond certain limits during the shelf life of the formulation (62). In this study, NE content (%) was determined as 90.00±0.01% on the day of production. This result shows that it is able to protect the YO loaded in the final NE formulation against degradation.

3.4. Morphology Analysis

Electron microscopy provides high-resolution images for structural examination of droplets nanoemulsion formulations. Figure 3 shows the TEM image of the YO-NE formulation. TEM analysis supported characterization studies of the synthesized nanoemulsion formulation. The image showed that the formulation had a spherical morphology and a homogeneous and monodisperse distribution, supporting the PdI results obtained from DLS analyses.

Figure 3: TEM image of YO-NE.

3.5. Accelerated Stability Tests

Stability is essential for formulations developed as product candidates. It provides information about the shelf life of the product candidate. A nanoemulsion formulation must remain physically stable with little or no change in droplet size throughout its shelf life. The creaming and phase separation rate of NEs (regarding shelf life) can be evaluated by gravitational force (63) and thermal stability (64). In the literature, emulsions have been assessed for any phase separation or turbidity as a result of centrifuge testing (27,60,65) and analyzed for organoleptic properties such as colour, odour, texture and phase separation as a result of thermal

stress testing (26,66). YO-NEs showed no physical changes or phase separation following 4,500 rpm gravitational force and thermal stability tests for 30 min (Table 2). In addition DLS results were given in Table 3.

3.6. Calibration Curve of YO

YO solutions were prepared in ethanol at different concentrations (2.34, 4.68, 9.36, 18.75, 37.5, 75 and 150 µg/mL). Then, the absorbance values of these samples were measured at 283.2 nm with a UV-Vis spectrophotometer, and the calibration curve shown in Figure 4 was drawn.

Table 2: Accelerated stability test data of the NEs.

Table 3: DLS results before and after preliminary stability tests.

 1.2 $v = 0.0071x + 0.0071$ $R^2 = 0.9995$ $\mathbf{1}$ Absorbance (283.2nm) 0.8 0.6 0.4

Figure 4: Calibration curve of YO (n=3).

3.7. Release Study Results of YO-NE

Oil-in-water (O/W) type NEs are unique systems that offer higher transparency, stability, biological activity, better physical and chemical properties, and better-controlled release (67). In this study, a release study of YO-NE formulation was carried out by comparing it with YO by diffusion membrane method under simulated skin pH and temperature conditions.

Figure 5 presents the % release of YO as a function of time. As seen in Figure 5, free YO was observed to be released from the membrane very quickly. While 98.84% of YO was released in the first 45 minutes, 99.98±1.00% of YO was released from YO-NE within 5 hours.

Figure 5: Release profile of YO-NEs.

While free YO molecules are released quickly, the release profile obtained by formulating YO as nanoemulsion ensures continuous release and longterm maintenance of a constant amount of oil (68,69).

The release of active ingredients from the nanoemulsion is controlled by interactions between the drug and surfactant or by partitioning the drug between the oil and water phases. Small droplet size and higher surface area in nanoemulsions will allow adequate release of the loaded drug. The release occurs in a controlled manner (70). Surfactant content is an essential factor affecting the release rate from the oil-containing nanoemulsion system. High surfactant ratio formulations have a slower release rate than low surfactant ratio formulations (71).

Shah et al (2019) (72) developed seven different NE formulations containing moxifloxacin, showing that these formulations released more than 90% of the drug content within 3 hours. Botros et al. (2020) (73) reported that almost all active ingredients were released from the formulation within 4.5 hours. These results are quite suitable for a topical preparation whose skin contact time is generally around 6 hours. Moreover, this short-term release indicates that its active ingredient is suitable for absorption and also provides good bioavailability (73). Our findings are similar to the literature.

3.8. Cytotoxicity Analysis of YO-NE

MTT assay was performed to determine the cytotoxic effect of YO and YO-NE formulation. This assay is

based on the metabolic activity of living cells by reducing MTT to formazan crystals. Since the overall mitochondrial activity of the cell population is related to the number of viable cells, the MTT assay is commonly used to evaluate the cytotoxic effects on cell lines or primary diseased tissues *in vitro* (74,75).

The cytotoxic activity of YO and YO-NE formulation was determined using HaCaT cells. HaCaT cells are easily used in skin toxicity studies due to being the first cell line subjected to cosmetic agents applied to the skin, high experimental reproducibility, ease of use, rapid proliferation and low cost (31). According to ISO standards, cell viability above 80% does not have a cytotoxic effect; 80% to 60% is low; 60% to 40% is medium; and below 40% is cytotoxic (76). The results showed that all concentrations of YO (0.25, 0.5, 1, 3, 5 mg/mL) reduced cell viability in HaCaT cells by 55.75 \pm 3.25 %, 41.10 \pm 3.64%, 40.70 \pm 3.65 %, 39.66 \pm 4.02%, and 31.34 \pm 2.04%, respectively (p<0.05). All concentrations of YO-NE (0.25, 0.5, 1, 3, 5 mg/mL) were found to have 96.62±1.44%, 83.29±2.52%, 82.30±2.64%, 81.69±1.24%, 67.95±0.80% cell viability in HaCaT cells, respectively (p<0.05). However, the concentrations of YO-NE tested (except 5 mg/mL: low cytotoxicity) did not show any cytotoxic effect on HaCaT cells. As shown in Figure 6, concentrations of 0.25 mg/mL, 0.5 mg/mL, 1 mg/mL, and 3 mg/mL YO-NE formulation had over 80% cell viability compared to the control group. This result indicates that the encapsulation of YO with NE makes it more cytocompatible compared to YO. Similar results were reported in literature (70,77,78).

Figure 6: Cytotoxicity data of YO and YO-NE on HaCaT cells. The asterisk (*) indicates that the difference between the control group and the treatment groups is significant at the p<0.05 level.

Solubility, instability and toxicity of lipophilic drugs pose problems. One of the most effective solutions to target these drugs is the formulation of NEs of lipophilic nutraceuticals. The solubility of lipophilic compounds increases as they disperse in the aqueous phase of the emulsion, preventing them from coming into direct contact with body fluids and tissues, thus reducing toxicity. Moreover, NEs provide high bioavailability of small amounts of encapsulated lipophilic substance to cells. They can, therefore, be used to study the uptake of encapsulated drug-active substance in cell cultures, improve the growth conditions and viability of cells, and conduct *in vitro* toxicity studies of lipophilic drugs (79). MTT results showed lower cytotoxicity of YO-NE compared to YO, indicating the nanoemulsion's controlled release properties and that YO's encapsulation is a suitable strategy to reduce its cytotoxicity towards HaCaT cells. Compared to YO, YO-NE still showed high biocompatibility even at higher concentrations and did not significantly affect HaCaT cell viability (80).

3.9. Analysis of Physiochemical and Pharmacokinetic Properties of Linalool and Germacrene-D

Kalagatur et al. (2018) (11) characterized ylangylang oil by GC-MS and reported that the main compounds in the oil's structure were linalool (29.15%), germacrene-D (11.82%), and thymol (8.45%). Physicochemical and pharmacokinetic analyses were performed to determine whether linalool and germacrene-D compounds in ylang-ylang oil are drugs (33). Various molecular properties such as absorption, distribution, metabolism, excretion, and structural properties such as molecular weight, number of hydrogen bond acceptors or donors, lipophilicity, and molar refractivity were investigated (39).

The physicochemical and pharmacokinetic properties of the compounds (Linalool, Germacrene-D) are

shown in Table 4 and Table 5. The Bioavailability Radar was displayed when predicting physicochemical properties such as lipophilicity, size, polarity, solubility, flexibility, and saturation (Figure 7). The radar plot shows that the selected compounds linalool and germacrene-D are within the pink area. This indicates a good bioavailability profile for the compounds and better drug similarity of the compounds alone. The pink area, the radar plot of molecules following Lipinski's rule of five, means that it is drug-like. According to Lipinski's rule of five, in order for a chemical compound to be orally active in humans, it must fulfill at least three of the following criteria: molecular weight \leq 500, XLOGP3 \lt 3.5, hydrogen bond acceptor ≤ 10 , hydrogen bond donor ≤ 5 and molar refractivity = 40-130 (36). In our study, Linalool complies with the rule of five. Germacrene-D violated the rule of one since MLOGP > 4.15. These two chemical compounds were accepted as drug analogs as drug as they met the Lipinski' rules. Therefore, ylang-ylang oil containing these chemical compounds has a high potential to be drug analogs (55).

The bioavailability of a drug depends on the absorption processes and first-pass metabolism in the liver, which depends on the solubility and permeability of the compound. Drugs with optimum log P and low molecular mass (<500) also have high permeability (81). In our study, these two compounds have high Caco-2 cell permeability as the permeability (nm/sec) > 0.90. The selected linalool and germacrene-D compounds exhibited negative Log Kp values (skin permeability) of -1.737 and - 1.429, indicating that they may be suitable as promising compounds for transdermal delivery.

James et al. (2023) (82) found the log Kp value of the linalool compound to be -1.43. Drioiche et al. (2023) (83) found the log Kp value of Germacrene-D compound to be -1.429. In estimating skin permeability, the log Kp value should be in the range

of -8.0 to -1.0. In this study, the Log Kp values of linalool and germacrene-D compounds were found in the recommended range and similar to the studies in

the literature. These results ensure good dermal penetration of these compounds.

Figure 7: Bioavailability radar showing and signaling drug susceptibility to Linalool **(a)** and Germacrene-D **(b)**

CYP450 and its isoforms belong to the hemoprotein family and play an important role in drug metabolism and clearance. Inhibition of isoforms leads to lower clearance and accumulation of the drug or its metabolism. Gastrointestinal (GI) absorption indicates the capacity of the drug to be absorbed and pass into the bloodstream. A compound without CYP450 inhibition and with high GI absorption properties implies that the compound has a good capacity for metabolization and absorption (39). In our study, Linalool and Germacrene-D compounds did not cause CYP450 inhibition and showed high GI

absorption of over 90%. The ability of a drug to cross the blood-brain barrier and enter the brain is a parameter that must be taken into account to help reduce or ameliorate side effects and toxicity (37). These two compounds provided values above log BB>0.3 and were found to be easily and well distributed into the brain and thus, able to bind to specific receptors. These two compounds showed good drug clearance values, and neither was a substrate for organic cation transporter 2 (OCT2). Furthermore, no hERG 1 channel inhibition or toxicity was predicted from the compounds.

3.10. Biological Activity Analysis of Linalool and Germacrene-D

The biological activity properties of Linalool and Germacrene-D compounds are presented in Table 6. Possible biological activities of Linalool and Germacrene-D compounds were obtained by the PASS server. Additionally, the set of pharmacological effects, mechanisms of action, and specific toxicities that may be exhibited by a compound in its interaction with biological entities were predicted by the PASS server (84). The software estimates the

predicted activity spectrum of a compound into probable activity (Pa) and probable inactivity (Pi). Only activities where Pa > Pi are considered possible for a given compound. If Pa > 0.7 , the probability of experimental pharmacological effect is high, if 0.5 < Pa < 0.7, the probability of experimental pharmacological effect is less, when Pa < 0.5, the chance of finding the activity experimentally is less, but it may indicate the chance of finding a new compound (40).

Table 6: Biological activities of Linalool and Germacrene.

Linalool contributes to topical anti-inflammatory and analgesic activities by inhibiting COX-2 protein expression in inflammatory tissues and reducing oxidative stress (85). Germacrene D reduces inflammation by preventing the release of inflammatory mediators by inactivating the enzymes that cause inflammation. One of the main mechanisms of Germacrene D is the inhibition of

prostaglandin synthesis. Prostaglandins are compounds that affect the inflammatory response and pain generation. This inhibition both alleviates inflammation and reduces pain generation (86). Linalool and Germacrene-D compounds play an essential role in the pathogenesis of inflammatory diseases. These compounds inhibit cell adhesion molecules involved in leukocyte traffic to provide

therapeutic intervention for leukocytes. With these properties, it is known to have therapeutic potential in human inflammatory disorders (87). These compounds exhibit various biological activities such as mucomembranous protector, cell adhesion molecule inhibitor, antisecretory, and antieczematic. It is understood that these compounds may be involved in the regulation of dermatological diseases caused by inflammation, with their remarkable antiinflammatory activities, as they have high pharmacological activity values (Pa > 0.7).

4. CONCLUSION

UV rays can cause various damage to the skin, such as erythema, pigment change, photoaging and skin cancer. Essential oils can contribute to the prevention or treatment of damage to the skin caused by UV rays due to their UV protective properties In our study, YO essential oil was used to protect against UV damage with its anti-inflammatory and antioxidant properties. YO was formulated in the NE dosage form to provide features such as increased skin permeability, bioavailability, solubility, therapeutic activity, stability, and controlled release. YO-NE was obtained with suitable average droplet size, PdI, and the ζ potential values, highly thermodynamically stable without sedimentation or phase separation, pH, and conductivity values suitable for topical application. It was determined that almost all the YO in the NE formulation was released within five hours. Cytotoxicity results showed that NE reduced the cytotoxicity of YO, and the YO-NE formulation had no toxicity. In conclusion, YO could be a formulation candidate that can be applied topically in treating dermatological diseases due to its controlled release feature, formulated in NE dosage form, non-toxicity, and the unique biological activities of Linalool and Germacrene-D compounds.

5. CONFLICT OF INTEREST

No conflict of interest.

6. ACKNOWLEDGMENTS

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